

# CHEMISTRY 4: CHANGES

## AND THE CONSERVATION OF MATTER:

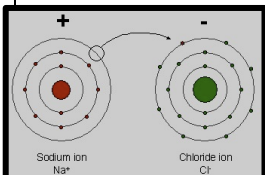
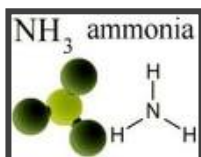
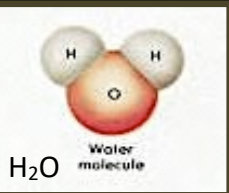
Matter (and Mass) is neither created nor destroyed, only rearranged, during normal chemical reactions.

SCIENCE 8 OCTOBER 2019 MRS. PLYTER

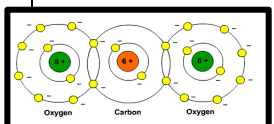
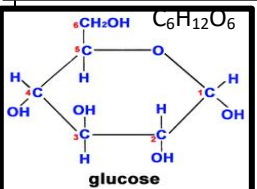
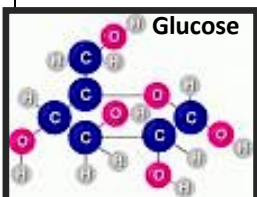
[HTTP://WWW.PLYTER.COM/SCIENCE](http://www.plyter.com/science)

Name \_\_\_\_\_

Period \_\_\_\_\_



NaCl: Table Salt



Carbon Dioxide CO<sub>2</sub>

# 15

**THE LAW OF CONSERVATION OF MATTER - MASS**  
Handout:

**A DIFFERENT ELECTRON DANCE**

# 16

**CHEMICAL CHANGE LAB**  
Endothermic or Exothermic?



# 17

**CHEMISTRY ONLINE**  
...or the iMac  
Get a Screen Check

Physical Science →  
Chemical Abbreviations  
Get a Screen Check \_\_\_\_\_

Classic ChemBalancer  
Get a Screen Check \_\_\_\_\_

Atom Builder Activity  
on the iMac Desktop  
Get a Screen Check \_\_\_\_\_

10 \_\_\_\_\_

Thu

# 18

IF TIME:  
**FIGURE IT OUT!**  
**WORKSHEETS**

Chemical Change Handout \_\_\_\_\_

Element and Compound Examples Worksheet \_\_\_\_\_

The Particle Model Worksheet \_\_\_\_\_

Fri

# 19

**CALENDAR DUE**

**Points:**  
Objective +Tape

Grade

The Law of Conservation of Matter

Chemical Change Lab

Online

Figure it out Worksheets

Daily Quizzes

[Write # Correct Initial in Color](#)

Mon \_\_\_\_\_

Tue \_\_\_\_\_

Wed \_\_\_\_\_

Thu \_\_\_\_\_

Fri \_\_\_\_\_

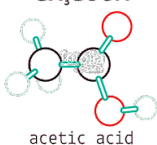
**TOTAL** \_\_\_\_\_

Photosynthesis

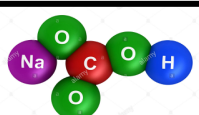


Carbon Dioxide + Water -----> Glucose + Oxygen

CH<sub>3</sub>COOH



Vinegar



Baking Soda

Central Science Home Page: [www.plyter.com/science](http://www.plyter.com/science)

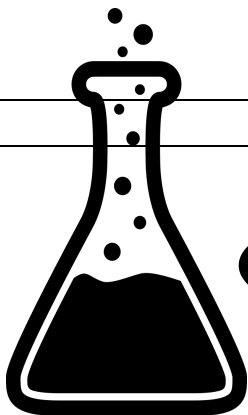
Physical Science: → Chemical Abbreviations  
Classic ChemBalancer

Discovery Education: (Username = yearlastf Student #)

Practice Tests: [MyGradebook.com](http://MyGradebook.com) Classword = plyter20 → Password = Your Student #

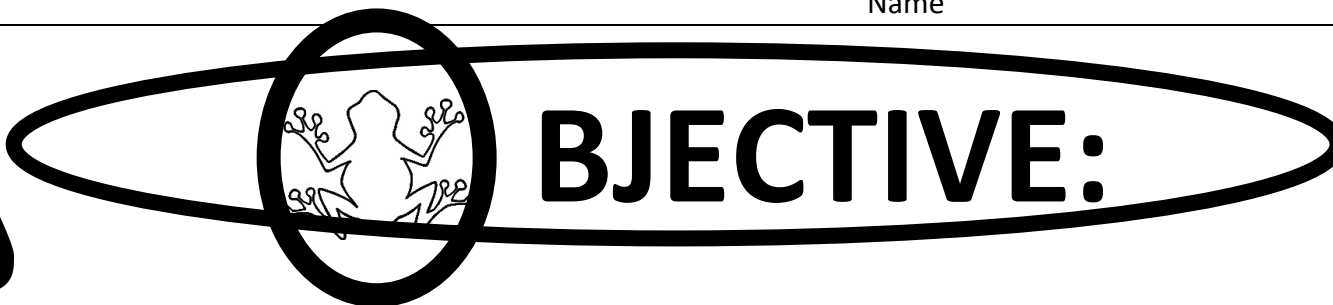
iMac: On the Desktop: Atom Builder Activity: Carbon





Name \_\_\_\_\_

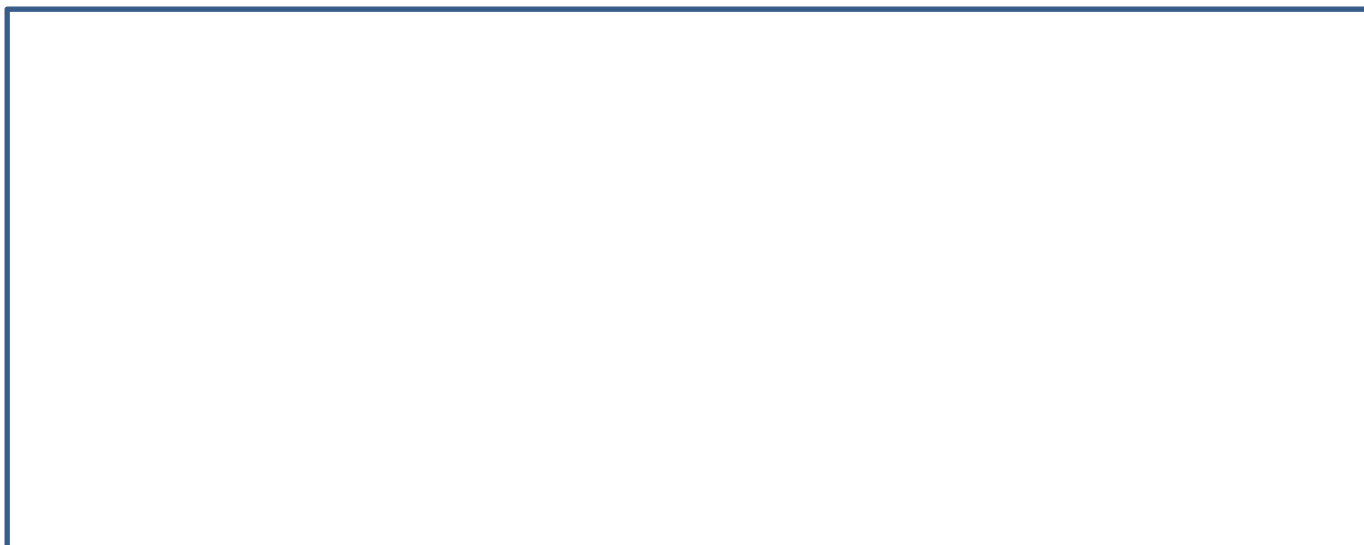
Period \_\_\_\_\_



Copy the Objectives: \_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_

**Electron Dance:**

- \_\_ 1) Place your **tape** in the rectangle.
- \_\_ 2) Illustrate one example of an Electron Dance using clear Tape.
  - \_\_ a) Write what you did to each tape.    \_\_ b) Draw results with -'s and +'s to show charges by each tape.



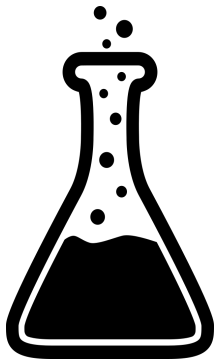
**Your Grade for Last Week:**

	<u>Yours</u>	<u>Required</u>
Calendar +	___	15
Grade	___	5
Investigate		
Background	___	10
Investigate	___	25
Ice to Steam	___	10
Try These	___	ex
Daily Quizzes	___	28
Totals	__	93

Yours / Required X 100 = %  
 \_\_\_\_\_ / 93 X 100 = \_\_\_\_\_

% Grade in Gradebook    \_\_\_

Last week my grade went  
 ↑? or ↓?    \_\_\_\_\_



- 1) Demonstrate energy transfer in a chemical reaction.
- 2) MODEL evidence for the Law of Conservation of Matter (mass) during chemical reactions by illustrating reactants and products.