

1. **Problem:** Develop a color chart for the pH INDICATOR, PURPLE CABBAGE JUICE.

2. **Background Information:**

- The pH of a solution indicates how A_____ IC a solution is.
- Measure pH with a substance called an IND _____ or a pH meter.
- The pH scale is a series of numbers from _____ to _____.
A pH of 7 means _____.
- A pH of less than 7 means an _____;
more than 7 means _____.
- The right amount of ACID added to a BASE in solution (or BASE to _____) will _____ ize it.
- A _____ is also called an ALKALINE or an ANTACID.
- Strong ACIDS have a pH value closer to _____;
weak ACIDS closer to _____.
- Strong BASES have a pH value closer to _____;
weak bases closer to _____.
- To clean up a STRONG ACID, you should use a WEAK _____,
such as _____.
- Examples: ACIDS: _____
BASES: _____
- pH INDICATORS are paper or liquid and change C_____.
- Each INDICATOR has its own C_____ and its own range of pH values.
- ACIDS are solutions that have extra _____ ions.
- BASES are solutions that have extra _____ ions.
- IONS have lost or gained ELECTRONS, so they have an electrical _____ and are attracted to others with an _____ electrical CHARGE.

3. Safety RULES: Initial each rule as you agree to it. Then SIGN.

- ___ 1. Work in your own lab. No more than 2 people to a group.
- ___ 2. Everyone records data and saves any proof needed.
- ___ 3. Equipment should be cleaned before use. DO NOT dry.
- ___ 4. Get solutions or substances as you need them.
DO NOT take Main Container. (continued) →
- ___ 5. DO NOT POUR BACK. If you get too much, ask.
- ___ 6. Goggles MUST BE WORN for pouring and mixing.
- ___ 7. LISTEN....READ....Follow directions and WARNINGS.
- ___ 8. CHEMICAL SOLUTION SPILLS must be rinsed with water before drying.

SIGN NAME here (BEFORE doing the LAB) : _____

4. Procedures: ✓ as you go.

- _1. Get a plastic stirring straw. Make a "tweezers" by cutting one end.
- _2. Obtain about 25 ml of Purple CABBAGE INDICATOR in a graduated cylinder. Store it in a clean beaker or other container.
- _3. Obtain 5 ml of a TEST SOLUTION in a clean graduated cylinder. RECORD (WRITE) the name of the solution in your chart.
- _4. Use a clean small plastic dish to MIX your 5 ml of TEST SOLUTION with 5 ml of Purple CABBAGE INDICATOR. STIR mixture with a clean plastic straw.
- _5. Obtain about 1 cm of pH indicator paper. Hold pH paper with your "tweezers" and dip it in your mixture. RECORD the first color and RECORD pH NUMBER from the colorchart.
- _6. ATTACH pH indicator paper for proof. (It will change color.)
- _7. Pour your mixture into a clean test tube.
Stand it in the test tube rack. Place a white paper behind it. Describe and RECORD the color of the mixture using 2 words.
- _8. RINSE WELL all containers and your stirring straw.
- _9. REPEAT Procedures #3-8, using a new TEST SOLUTION
- _10. If you run out of time...or finish
Have your results CHECKED BEFORE YOU CLEAN UP. →

