

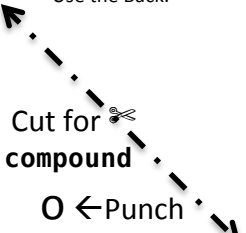

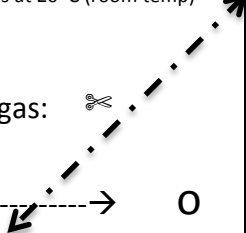


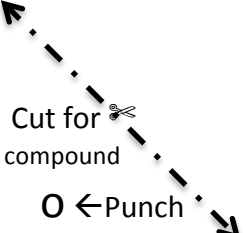

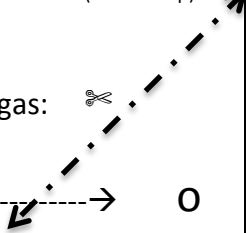


Compound Card

<p>1) Molecular Mass: amu (atomic mass unit): Round to the nearest whole number. Element amu X # atoms = subtotal _____ _____ _____ Total = Molecular Mass _____</p>	<p>2) Your Name: _____ Period: _____</p>	<p>3) Formula: Write LARGE. Watch UPPER and lower case. Use subscripts correctly _____ Compound Name: _____ Common Name: _____</p>
<p>4) Structure: _____ p⁺ = protons+ _____ (Atomic #) n⁰ = neutrons⁰ _____ (p⁺ plus n⁰ = amu) e⁻ = electrons- _____ (e⁻ equals p⁺) Electron Levels: _____</p>	<p>5) Physical Properties: Density _____ So, in water it _____ Boiling Point (pt): _____ °C Melting/Freezing Pt _____ °C So, at room temperature, it is a _____</p>	<p>6) Atomic Numbers Element At# _____ _____</p>
<p>7) Diagram: Use the Back.  Cut for compound  O ← Punch </p>	<p>8) Fact: Use: Source:</p>	<p>9) State of Matter: Solid, Liquid or Gas at 20° C (room temp) Cut for gas:  Punch-----→ O </p>

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