Diffusion and Osmosis Lab page 1 of 2 Name	Period
A. Class Lab:a) Those at table number 1-4 = salt; those at all other tables = waterb) A "cell membrane" divides room between tables 4-5-6 and 7-8-9c) Check the board, or ask, "When?". MODEL osmosis; then diffusiond) Have another student sign the you participated.	
Work with another student. Agree as you go $\downarrow$ Diffusion	? 6
B. Diffusion is the movement of molecules from a region of higher concentration to a region of lower conce	A A A A A A A A A A A A A A A A A A A
1. <u>Color</u> the DIFFUSING molecules of <u>sugar</u> . $\rightarrow$	
2. Diffusion happens because of random molecular motion. Molecules m randomly until there is an even mixture throughout their container. Movement container (the container could be a cell or a room) occurs until the mixture is even throughout the container. An example is perfume out in a room.	within a
con	ectively Permeable Membrane
4. Cell membranes do not allow ALL molecules to cross them. Membranes are be semipermeable, "selectively" or "differentially" permeable. Your filter paper "selective" as it let water and s go through, but not the s 5. For example, some semipermeable membranes allow water but not dissolve (sodium and chlorine ions) to pass through. If salt is more concentrated on one than on the other, the water will move across the membrane to the salt side make things more even or equal. The salt cannot get through, so salt has to so movement process is called osmosis. In the above drawing the moved the mode the water level as	r was ed salt side <b>to</b> tay. The
the m to the right, and made the water level go 6. Food, water, Ox, Carbon D, and other wastes DIFFUS your body. When they cross a cell, it is OSMOSIS	E around
Have signed by another student that agrees with your answers	

Diffusion and Osmosis Lab page 2 of 2 Name	Period
Do your own labs.	ů_
D. Osmosis Lab: Osmosis through cell membranes.	RED
1. Obtain 3 test tubes & 3 plant pieces that will fit into the tubes. See	chart.
$\underline{\hspace{0.5cm}}$ 2. Obtain about $\frac{1}{2}$ tube of each liquid for osmosis. Label or keep in order	r in a holder.
3. Measure mass(g) of each plant piece. Record in order.	
4. Add pieces, in order, as recorded in your chart. The liquid should cover	er each.
5. Wait. If overnight, label. Include your name and period; place in corn	rect tray.
6. If time, do the Diffusion Lab below.	
7. After minutes or overnight, measure mass again. Record any ob	servations.
8. Explain your results telling what moved and using the word Osmosis.	
9. Clean up. Plant material goes in the trash.	
E. Diffusion Lab:	
10. Obtain 3 test tubes and a beaker or other holder. See Diffusion Cha	art below.
11. Obtain close to the same amount of each of the three liquids for diff	fusion,
leaving some room at the top of the test tube for color.	RO.
12. Be ready to record the time and have all of your tubes together.	
13. Take all tubes to the pump and quickly add one "pump" of colored wat	er to
each. Record time. Do not stir or shake. Place in your beaker. Ob	oserve.
14. Record time again, for each, when the color reaches the bottom of t	he tube.
If out of time, record the time and the progress of the color.	

Diffusion	Cold	Hot	Salt	Osmosis:	City	Salt	Distilled
of Color	Water	Water	Water	Plant Cells	Water	Water	Water
Start time				Before: mass(g)	)		
End Time				After mass(g)			
Progress				Changes			
Analysis: Use the word diffusion in a complete, descriptive, sentence that explains. :			Analysis Use the word osmosis in a complete, descriptive, sentence that explains.				